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Marginal abatement cost analysis of cleaner power production alternatives for sustainable energy transition

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Energy transition is a global transformation driven by greenhouse gas climate change in which fossil power production is replaced by fully renewable counterparts. Currently, the world has accomplished a small fraction of the energy transition. The main reason is that fossil energy sources are still abundant, cost-effective, portable, and easily implementable, while renewable sources are still technologically incipient, costly, cumbersome, portable, intermittent/seasonal, and land-intensive. Because the world cannot suddenly get rid of fossil power production, the compromise is to convert it into sustainable production by attaching carbon abatement processes (anti-carbon packages). Moreover, it is also conducive to migrating from carbon-intensive fossil power production to less carbon-intensive ones. This trend is perceptible as coal-fired plants are being replaced by natural gas combined cycles characterized by maximum thermodynamic yields and minimum carbon emitted per kWh among all fossil analogs. Thus, a conceivable next step of energy transition is to implement natural gas combined cycles with anti-carbon packages. This work compares three anti-carbon packages—post-combustion, oxy-combustion, and pre-combustion—attached to a typical natural gas combined cycle. A differential is the novel marginal abatement cost (MAC) adopted as an economic-environmental performance metric. It is shown that the post-combustion combined cycle requires the minimum investment (596.89 MMUSD) and attains the maximum net value (2,060.68 MMUSD) and minimum MAC (4.58 USD/tCO2), making it the best economic-environmental compromise. The oxy-combustion combined cycle attains the maximum investment (1,095.19 MMUSD) and intermediary MAC (12.09 USD/t^{CO2}) and net value (1,144.21 MMUSD) but offers the best social benefit per environmental harm ($Power^{Exported}/CO_2^{Emitted} = 51.63 MWh/t^{CO2}$). The pre-combustion combined cycle has a minimum net value (339.66 MMUSD) and a maximum MAC (28.17 USD/t^{CO2}) and is likely the worst decarbonized combined cycle according to all metrics.

KEYWORDS

cleaner energy, sustainable energy, power production, marginal abatement cost, post-combustion, oxy-combustion

1 Introduction

Energy transition is a global technological transformation driven by greenhouse gas climate change, in which fossil power production is completely replaced by fully renewable counterparts. Nowadays, it is well recognized that the world has accomplished only a small fraction of the necessary energy transition (Ahmad and Zhang, 2020). The underlying reason is that fossil energy sources are still abundant, affordable, portable, reliable, and easily implementable, while renewable sources are still technologically incipient, costly, cumbersome, portable, intermittent/seasonal, and land-intensive and compete with forest and food-producing lands. Because the world cannot suddenly get rid of fossil power production, a feasible and necessary compromise is to convert ordinary fossil power production into sustainable production by attaching carbon abatement processes or, as denominated here, anti-carbon packages. Sustainable fossil power production still could inject fossil carbon into the biosphere, but the anti-carbon packages capture a good portion of it and eliminate it as geological sequestration (carbon capture and sequestration (CCS)) or to some industrial manufacture of durable commodities using this carbon (carbon capture and utilization (CCU)). Many countries have implemented mechanisms such as carbon pricing, emissions trading systems, and taxation policies that assign a financial cost to carbon emissions, encouraging industries to adopt cleaner practices and invest in low-carbon technologies. In response, CCS and CCU technologies have emerged as critical components in the strategy to develop sustainable fossil power production without compromising the energy transition (Araújo and de Medeiros, 2021).

It is also helpful to implement technology changes that result in replacing carbon-intensive fossil power production with less carbon-intensive forms. This trend is perceptible as ordinary coalfired plants are gradually being replaced by integrated gasification combined cycle (IGCC) and/or natural gas (NG) combined cycle (NGCC) plants (Ahmad and Zhang, 2020). IGCCs are very capitalintensive and have greater heat intake, while NGCCs are less capital/heat-intensive and attain the highest thermodynamic yields, entailing minimum carbon emitted per kWh among all fossil fuel analogs (Bhattacharya and Datta, 2013). Even so, NGCCs continue to emit considerable amounts of carbon dioxide (CO2), demanding integration into anti-carbon packages. Consequently, a conceivable next step in the energy transition is to encourage the implementation of NGCCs with anti-carbon packages. The marginal abatement cost (MAC) is a widely accepted metric (expressed in USD/t^{CO2}) used to measure the economic-environmental performance of sustainable fossil power production (and also renewable power production) (Lameh et al., 2022).

Regarding the anti-carbon packages for sustainable fossil power production, three kinds of technology exist: (i) post-combustion carbon capture, (ii) oxy-combustion carbon capture, and (iii) precombustion carbon capture.

Post-combustion capture is easily integrated into existing power plant facilities, as it is mainly located downstream, where CO₂ is removed from flue gas. This separation is typically achieved through either absorption or adsorption, with aqueous ethanolamine absorption being the most widely adopted technology (Goren et al., 2024). Regarding adsorption technologies, several

developments exist. Calvo-Muñoz et al. (2016) created post-combustion adsorbents from biomass waste, and Bharath and Rajendran (2024) developed a temperature-swing adsorption (TSA) technology for CO₂ removal from NGCC flue gas with steam-assisted adsorbent regeneration. Isogai and Nakagaki (2024) economically assessed an NGCC with post-combustion capture via aqueous monoethanolamine (PCC-MEA) absorption, in which the heat for solvent regeneration shifts to power-to-heat during low electricity demand periods. Cruz et al. (2023) applied PCC-MEA to NGCC or to single-cycle power generation in offshore rigs, and Yadav et al. (2025) applied post-combustion capture to a NGCC flue gas using cold energy from liquefied NG regasification.

In oxy-combustion carbon capture, fossil fuel is burned with oxygen (O2 > 96% mol/mol), resulting in a flue gas composed primarily of CO₂ and water vapor. This composition enables straightforward CO₂ separation, requiring only water condensation. An air separation unit (ASU) is required to supply the necessary oxygen. ASU technologies comprise cryogenic distillation, pressureswing adsorption, and membrane permeation. An alternative oxycombustion approach is the chemical-looping combustion, although this technology is still in development (Goren et al., 2024). Among the available ASU options, cryogenic distillation is the only one commercially established at a high capacity that has been extensively studied. In this context, Brigagão et al. (2019) studied a new cryogenic ASU based on top-vapor recompression distillation with lower power consumption integrated with an oxycombustion NGCC with CCU. Skorek-Osikowska et al. (2015) presented an economic and thermodynamic comparison between two ASU concepts for oxygen supply to a coal-fired oxycombustion power plant: conventional cryogenic distillation and a hybrid process combining membrane separation with cryogenic distillation. Tan et al. (2023) studied biomass gasification integrated with oxy-combustion NGCC fed with liquefied NG as a strategy to reduce energy consumption through the utilization of its cold energy. Khallaghi et al. (2019) studied a staged oxycombustion NGCC to avoid exhaust-gas recirculation (EGR) and used supercritical CO_2 as the working fluid of the bottoming cycle.

In pre-combustion capture systems, high-pressure synthesis gas (syngas), containing hydrogen (H2), carbon monoxide (CO), and CO2, is produced from a fossil fuel via steam reforming or steam gasification. Syngas becomes shifted syngas-H2 + CO2-after a water-gas shift (WGS) reaction, and the shifted syngas is decarbonated before power production, in which H2 is burned with air (Malekli et al., 2025). Compared to post-combustion technologies, the high CO2 fugacity in the high-pressure shifted syngas enables easy CO2 absorption, allowing the use of less heat-intensive solvents instead of aqueous MEA (Geweda et al., 2025). In addition to the high thermal loads required in steam reforming or steam gasification, retrofitting ordinary power plants to a pre-combustion configuration entails extra high investment and complexity (Singla et al., 2025). Moreover, the WGS reactors impose some process inefficiency (Thiedemann and Wark, 2025). Several works have explored the context of NGCC with precombustion capture. Cobden et al. (2007) studied a sorptionenhanced hydrogen production with hydrotalcite-based materials for high-temperature adsorption of CO₂ (≈400 °C). Solares and Wood (2020) developed a scale-up model for pressure-swing

adsorption for CO2 removal from shifted syngas. Pruvost et al. (2022) conducted a techno-economic assessment of blue, green, and turquoise pathways for hydrogen production from NG, comprising pre-reforming, steam-methane reforming (SMR), WGS, and pressure-swing adsorption. Oh et al. (2022) studied bluehydrogen production from NG with a process comprising SMR without WGS afterward, syngas cooling (without heat recovery), and CO₂ removal via absorption by aqueous methyl-diethanolamine (MDEA) with piperazine (aqueous MDEA-PZ). Zanetti et al. (2025) studied an offshore floating plant for NG conversion into blue-hydrogen wherein CO2 is monetized as a fluid for enhanced oil recovery (EOR) in offshore oil-gas fields. Membrane separation technologies have also been considered for shifted syngas decarbonation, namely, (i) Belaissaoui and Favre (2018) studied a gas-liquid membrane contactor with Selexol and (ii) Ni et al. (2025) contemplated membrane permeation exploring the leverage of high CO₂ fugacity.

Some works compared post-combustion, oxy-combustion, and pre-combustion in the context of NGCC with CCS/CCU. Kanniche et al. (2010) evaluated these three anti-carbon packages for different sorts of power plants, and the results suggest the following best matches: (i) pre-combustion capture with physical-solvent absorption for integrated gasification combined cycle; (ii) oxy-combustion for coal-fired power plants; (iii) post-combustion capture via aqueous MEA absorption for NGCC. Kazemi et al. (2022) performed a techno-economic assessment of 19 alternative NGCC configurations with post-combustion, oxy-combustion, and pre-combustion. Kazemi et al. (2023) performed economic optimization and comparative environmental assessment of 11 NGCC configurations.

Table 1 summarizes the literature reviewed on anti-carbon packages for sustainable fossil power production.

An inherent characteristic of all studied NGCC power plants with anti-carbon packages discussed above has to do with the fact that none of them considered the evaluation of economic-environmental performance via the marginal abatement cost (MAC). MAC is currently considered a better and concise metric for the evaluation of sustainable power production processes. This work compares the three kinds of anti-carbon packages—post-combustion, oxy-combustion, and pre-combustion—attached to typical NGCC plants using the novel MAC as a metric (in addition to other classical metrics like the net present value and ratios, such as MWh^{Exported}/t^{CO2-Capt}). It is shown that the three anti-carbon packages have typical signatures (in terms of MAC and other metrics) when coupled to NGCC power plants.

2 Materials and methods

A typical NGCC power plant was chosen as the base case and designed first. Then, its post-combustion, pre-combustion, and oxy-combustion variants were formulated, simulated, and designed using the Aspen HYSYS simulator. All the necessary pertinent information and technical resources (e.g., flowsheets) are presented throughout this section. Other methods for economic analysis and *MAC* analysis are also discussed.

2.1 Process description

The four analyzed processes: base case, post-combustion variant, pre-combustion variant, and oxy-combustion variant, are described through a set of six subsystems that are combined to form them. The elemental subsystems are (i) a gas turbine block (GT); (ii) a Rankine cycle block (RANK); (iii) post-combustion capture via an aqueous MEA absorption block (PCC-MEA); (iv) a CO₂ compression train and dehydration with a triethylene-glycol block (CC-TEG); (v) a cryogenic air separation unit block (ASU); and (vi) steam reforming and WGS for a H₂ production block (HU). Figures 1-3 describe the flowsheets of the six elemental subsystems. Certain subsystems, such as GT and RANK, are defined as super-structures (i.e., with extra streams) to allow all respective possible operational modes. The six subsystems are described in Sections 2.1.1-2.1.6. The four analyzed processes are represented as block diagrams in Figures 4, 5 and are discussed in Sections 2.1.7, 2.1.8. The four processes include a cooling-water (CW) tower, which is omitted in the respective diagrams.

2.1.1 Gas turbine block (GT)

A GT (Figure 1a) consists of an axial adiabatic air (and/or flue gas) compressor, an adiabatic combustor, and an axial adiabatic expander. To simulate GT operation, compressor and expander adiabatic efficiencies were previously calibrated to match the performance data of the selected commercial gas turbine. Ambient air and/or flue gas recirculation (from RANK) feed the compressor to enable high-pressure combustion of fuel gas (NG or H2) with air (or oxygen). Air excess and/or flue gas recirculation flow rates are regulated to maintain the maximum allowable combustion temperature. However, increasing the total air (or flue gas) inlet flow rate increases compressor shaft work. High-pressure fuel gas (NG or H₂) directly feeds the combustor, and the resulting hightemperature, high-pressure flue gas expands through the axial expander. The compressor, expander, and generator share a single shaft, such that the expander drives the compressor and the electricity generator. The flue gas from the expander is nearly atmospheric but still has significant enthalpy at high temperature.

2.1.2 Rankine cycle block (RANK)

In the RANK (Figure 3b), the heat-recovery steam generator (HRSG) transfers enthalpic content of hot (nearly atmospheric) flue gas (FGATM for oxy-combustion, post-combustion, and precombustion, and FGH2 also for pre-combustion) to produce high-pressure steam (HPS) from high-pressure condensate and low-pressure steam (LPS) from low-pressure condensate (LPS production is disabled in the base case). HPS expands through a steam turbine that drives another generator, complementing the electricity generated in the GT. The steam turbine exhaust condenses in the Rankine cycle condenser and is pumped back to the HRSG, closing the Rankine cycle. The final flue gas (FGATM) is further cooled in a direct-contact column (DCC), and a part of it becomes the exhaust-gas recirculation (EGR) to the GT in the postcombustion and oxy-combustion processes. The final flue gas FGH2 only exists in the pre-combustion NGCC, and it is cooled in a shelland-tube exchanger to recover demineralized water condensate to feed the hydrogen unit (HU) as process water for steam generation (used in SMR).

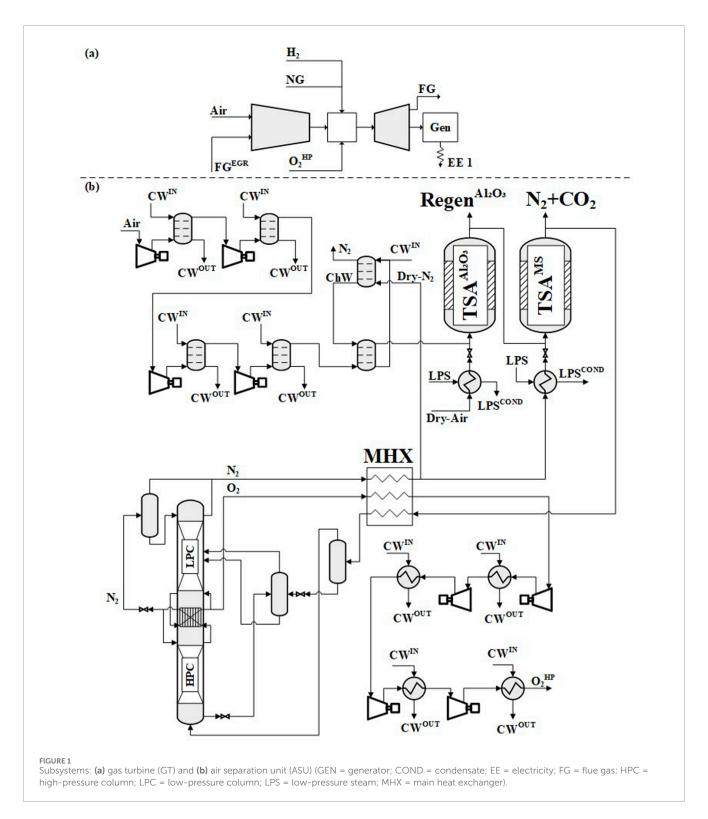
TABLE 1 Comparison of literature on anti-carbon packages for sustainable fossil power production.

Reference	Anti-carbon package	Description		
lvo-Muñoz et al. (2016) Post-combustion capture		NGCC with flue gas decarbonation via adsorbents from biomass waste. Steam-assisted adsorbent regeneration		
Bharath and Rajendran (2024)	Post-combustion capture	NGCC with flue gas decarbonation via TSA with steam-assisted adsorbent regeneration		
Isogai and Nakagaki (2024)	Post-combustion capture	NGCC with flue gas decarbonation via PCC-MEA: solvent regeneration via power-to-heat for low electricity demand period:		
Cruz et al. (2023)	Post-combustion capture	NGCC or single-cycle power generation with flue gas decarbonation via PCC-MEA. Offshore rigs		
Yadav et al. (2025)	Post-combustion capture	NGCC with flue gas decarbonation. Employs cold energy from liquefied natural gas regasification		
Brigagão et al.	Oxy-combustion	NGCC with CCU. ASU via top-vapor recompression distillation		
Skorek-Osikowska et al. (2015)	Oxy-combustion	Two ASU concepts: (i) conventional cryogenic distillation; (ii) membrane separation with cryogenic distillation. Economic and thermodynamic comparison		
Tan et al. (2023)	Oxy-combustion	Biomass gasification integrated with oxy-combustion NGCC with liquefied NG; lower energy consumption		
Khallaghi et al. (2019)	Oxy-combustion	Staged oxy-combustion NGCC to avoid EGR. Supercritical ${\rm CO}_2$: working fluid of the bottoming cycle		
Cobden et al. (2007)	Pre-combustion capture	Hydrotal cite-based materials for sorption-enhanced CO_2 adsorption from shifted syng as		
Solares and Wood (2020)	Pre-combustion capture	A scale-up model for pressure-swing adsorption for carbon capture from shifted syngas		
Pruvost et al. (2022)	Pre-combustion capture	Techno-economic assessment of blue, green, and turquoise pathways for hydrogen production		
Oh et al. (2022)	Pre-combustion capture	Blue-hydrogen production: SMR without WGS, syngas cooling without heat recovery, and $\rm CO_2$ removal with aqueous MDEA-PZ		
Zanetti et al. (2025)	Pre-combustion capture	Blue-hydrogen production in an offshore floating plant. ${\rm CO_2}$ monetization via EOR in offshore oil-gas fields		
Belaissaoui and Favre (2018)	Pre-combustion capture	Shifted syngas decarbonation via gas–liquid membrane contactors with Selexol		
Ni et al. (2025)	Pre-combustion capture	Shifted syngas decarbonation via membrane permeation		
Kanniche et al. (2010)	Post-combustion, oxy-combustion, and pre-combustion	Evaluated three sorts of anti-carbon packages for different sorts of power plants		
Kazemi et al. (2022)	Post-combustion, oxy-combustion, and pre-combustion	Techno-economic assessment of 19 NGCC power plant configurations		
Kazemi et al. (2023)	Post-combustion, oxy-combustion, and pre-combustion	Economic optimization and comparative environmental assessment of 11 NGCC power plant alternatives		

Because implementation of all anti-carbon packages requires LPS consumption, RANK also produces LPS in the HRSG, reducing HPS production and, consequently, electricity generation in the steam turbine. In other words, decarbonization implementation entails reducing NGCC net exported electricity (this loss is part of the so-called capture penalty).

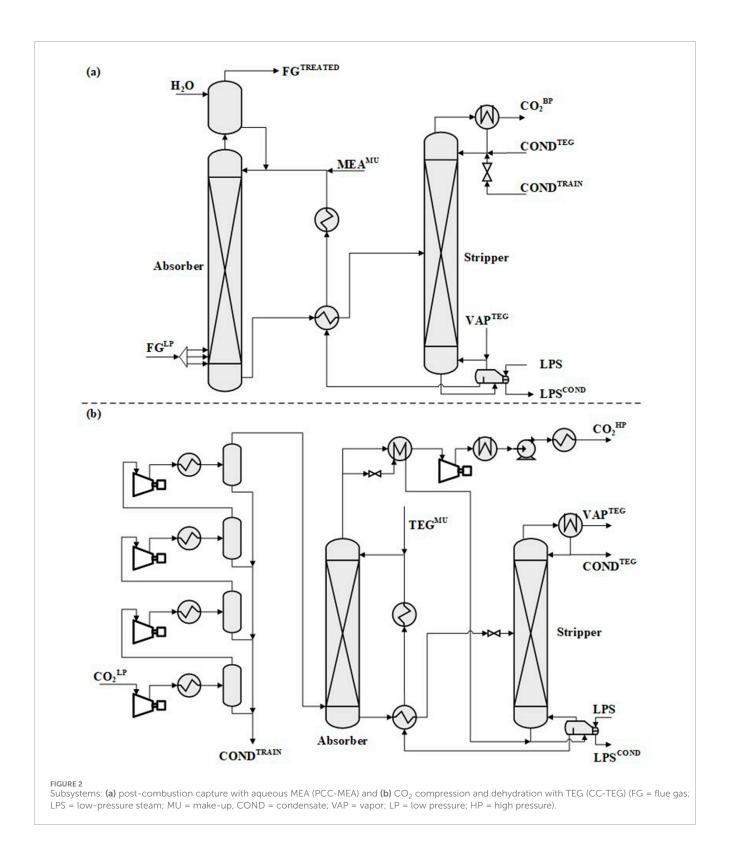
2.1.3 Post-combustion capture via aqueous MEA absorption block (PCC-MEA)

In PCC-MEA (Figure 2a), the separation of CO_2 from flue gas, composed of CO_2 , N_2 , argon, O_2 , and water, is achieved through chemical absorption with aqueous MEA. Flue gas feeds the absorption column at atmospheric pressure and is uniformly



distributed among the bottom stages to enhance separation efficiency. Lean aqueous MEA is fed at the top and absorbs CO_2 via reversible exothermic reactions. The absorber bottom product is CO_2 -rich aqueous MEA, while the water-saturated top vapor contains decarbonized flue gas, water vapor, and traces of MEA. To minimize solvent losses, the top vapor is cooled in an external vessel, where more than 99% of the MEA is condensed and recovered. CO_2 -rich aqueous MEA is pre-heated with hot lean

aqueous MEA and feeds the regeneration stripper column, wherein CO_2 is stripped from the solvent. Additional feeds to the stripper include carbonated waters fed at Stage#1 and a water-saturated CO_2 stream injected in the stripper reboiler, namely, (i) aqueous condensates from the CO_2 compression train; (ii) aqueous distillate from the TEG dehydration unit; and (iii) water-saturated CO_2 vapor from the TEG unit. These carbonated waters reduce condenser heat duty, water make-up, and CO_2 fugitive emissions, while the



water-saturated CO_2 feed also reduces water make-up and CO_2 fugitive emissions and lowers the reboiler temperature, helping prevent MEA degradation. The lean aqueous MEA is cooled and pumped back to the absorber along with recovered liquid and make-up solvent, maintaining solvent composition for continuous operation.

2.1.4 Carbon dioxide compression and dehydration block (CC-TEG)

Captured CO_2 is water saturated and must be dehydrated after compression to prevent CO_2 hydrate formation in the pipeline, a phenomenon favored by high pressures and temperatures below 10 °C. CC-TEG (Figure 2b) executes initial CO_2 compression and

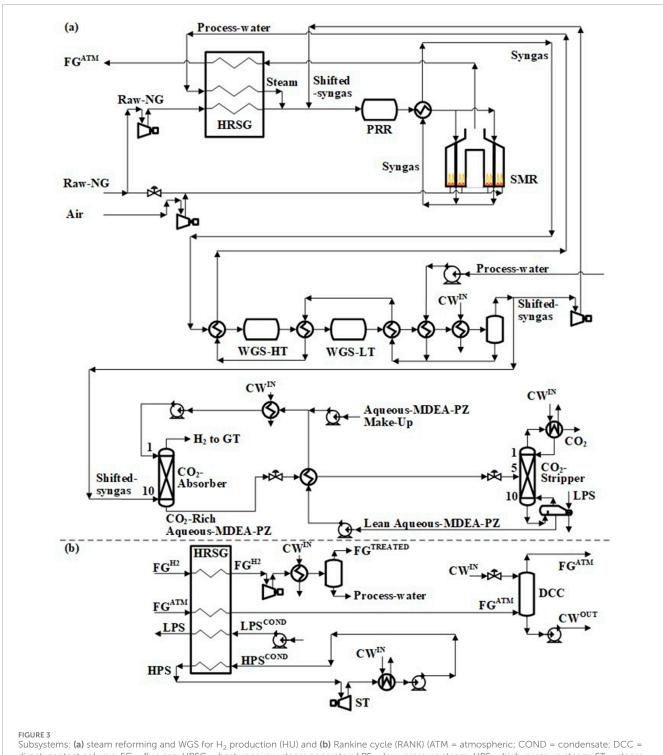
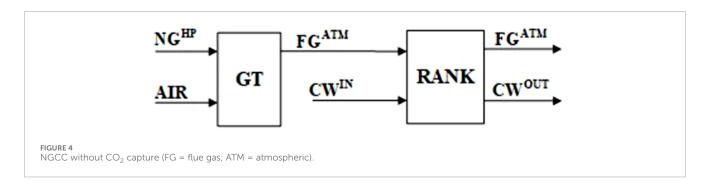
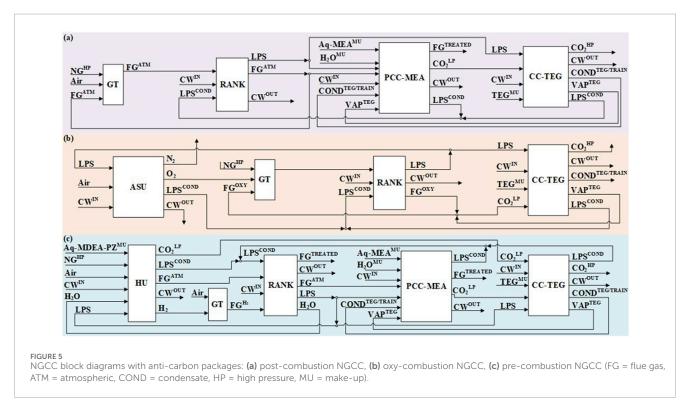


FIGURE 3
Subsystems: (a) steam reforming and WGS for H_2 production (HU) and (b) Rankine cycle (RANK) (ATM = atmospheric; COND = condensate; DCC = direct-contact column; FG = flue gas; HRSG = heat-recovery steam generator; LPS = low-pressure steam; HPS = high-pressure steam; ST = steam turbine; LT = low-T, HT = high-T; PRR = pre-reforming reactor; SMR = steam-methane reforming).

dehydration via TEG absorption at $\approx 65\ bar$. Dry CO $_2$ is compressed once more and pumped after condensation for pipeline dispatch. Figure 2b (West) depicts four compression stages followed by intercoolers and knock-out vessels for condensed water removal. Train power consumption is minimized, setting equal stage compression ratios (≈ 3). Condensed waters from knock-out vessels

contain CO_2 and return to the PCC-MEA stripper to reduce fugitive emissions and water make-up. After the 4-stage compression train, CO_2 is fed at the bottom of the TEG absorber, while lean TEG is fed at the top. High-pressure CO_2 dehydration has some advantages: (i) increases TEG-water affinity and equilibrium water loading; (ii) reduces column diameter by increased CO_2 density; (iii) reduces





water absorption load due to partial water condensation in the compression train intercoolers; and (iv) reduces TEG evaporative losses into dry CO₂. However, the dehydration pressure should not surpass $P \approx 70 \, bar$ due to CO₂ supercritical effects that stabilize water vapor, hindering dehydration. A small fraction (≈2%) of dry CO2 is deviated, expanded, and injected into the TEG stripper reboiler as stripping gas. This enhances water stripping, reducing the reboiler temperature (limited to $T = 140 \,^{\circ}\text{C}$ to avoid TEG degradation). Simple measures can extend TEG durability up to three or more years (Zanetti et al., 2025), such as (i) injecting dry stripping gas (no water and no oxygen allowed) to keep the reboiler temperature below T = 140 °C; (ii) avoiding air entrance; and (iii) adding an oxidant scavenger (i.e., 0.1% w/w NaHSO₃). Both vapor and aqueous liquid distillates from the TEG stripper recycle to the PCC-MEA stripper to minimize CO2 fugitive emissions and water losses. In the present case, simulation shows that CC-TEG recovers approximately 15.7 t/h of water and avoids 10.2 t/h CO₂ fugitive emissions. These measures improve the economic and environmental sustainability of CC-TEG. Dry CO2 is compressed again from $P \approx 65\ bar$ to $P \approx 100\ bar$ and cooled in the intercooler to $40\ ^{\circ}\text{C}$. At $P \approx 100\ bar$ and $T = 40\ ^{\circ}\text{C}$, CO₂ condenses as a dense supercritical fluid. Dense CO₂ is pumped to $P = 250\ bar$ via a supercritical-fluid pump, cools to $T = 40\ ^{\circ}\text{C}$, and is then dispatched to the pipeline. High pressure and low temperature increase the fluid density, reducing the pipeline diameter and/or fluid velocity, and lowering the head loss per kilometer.

2.1.5 Cryogenic air separation unit (ASU)

The ASU subsystem (Figure 1B) adopts the cryogenic double-column Linde process for air fractionation and is designed to supply high-purity oxygen to NG oxy-combustion. High-purity nitrogen (N_2) is also produced. The air feed undergoes four compression stages in adiabatic centrifugal compressors. Air from each stage is cooled in an interstage DCC with CW, which also partially dehydrates air due to the increasing pressure along the train. A final cooling step uses chilled water (ChW) in another DCC, further reducing air temperature and its humidity prior to its absolute dehydration via TSA with an activated alumina bed. ChW is

produced via evaporative DCC, where dry N_2 at $T \approx 15$ °C from the cold box contacts CW (T = 35 °C), promoting water evaporation and chilling CW to ChW. Before entering the cold box, the air must be absolutely dehydrated and decarbonated to prevent ice and dry-ice blockages in the cryogenic equipment. Water and CO₂ are, respectively, removed through TSA with an activated alumina bed and another TSA with a 13X zeolite bed. The alumina bed is regenerated with a fraction of dry air heated to 200 °C, while the zeolite bed is regenerated with the remaining dry N₂ from the cold box heated to T = 120 °C. After dehydration and decarbonation, the air is sufficiently purified for cold box conditions. At the cold box entrance, purified air is cooled in the main heat exchanger (MHX) with cryogenic liquid O₂ and N₂ vapor leaving the cold box. In the MHX, air attains cryogenic temperatures and partially condenses, wherein its liquid and vapor phases are separated in a cryogenic high-pressure flash. The nitrogen-rich vapor feeds the bottom of the high-pressure column (HPC), while the oxygen-rich liquid expands and feeds the low-pressure column (LPC). The pressure difference between the HPC and the LPC must guarantee that the condensing temperature of N₂ at the top of the HPC remains higher than the boiling temperature of oxygen at the bottom of the LPC. Oxygenrich liquid is collected at the bottom of the LPC, while a nitrogenrich vapor is a product at the top of the LPC. These products leave the cold box through the MHX. Outside the cold box, the gaseous O₂ product passes through a 4-stage intercooled compressor train to reach GT combustor pressure (Figure 1a). Direct injection of O₂ into the GT combustor prevents risks of combustion/explosion in the GT axial compressor of the oxy-combustion plant due to potential contact with residual hydrocarbons in the EGR fluid.

2.1.6 Hydrogen unit (HU)

The HU subsystem (Figure 3a) is fed with NG divided into two sub-feeds: (i) 55% as a fuel gas to heat steam-methane reforming (SMR); (ii) 45% as a reactant for the SMR reaction ($P = 28.35 \ bar$). Fuel gas depressurizes to $P = 1.693 \ bar$, mixes with air (50% excess), and is burned in SMR combustors. Hot flue gas (FG^{ATM}) goes to RANK after releasing heat at $T \approx 975 \ ^{\circ}$ C in the HRSG (Figure 3a) for (i) NG pre-heating to $T = 390 \ ^{\circ}$ C as SMR reactant and (ii) superheated steam generation ($T = 500 \ ^{\circ}$ C).

In the SMR section, NG and steam are mixed $(mol^{H2O}/mol^C=3)$ with some shifted syngas to adjust $mol^{H2}/mol^C=0.5$ to avoid coke formation in pre-reforming. Pre-reforming converts C_2^+ hydrocarbons into syngas $(H_2, CO, and CO_2)$ via Equations 1–3 to avoid coke formation in SMR (Zanetti et al., 2025). Pre-reforming effluent is pre-heated to 540 °C with SMR syngas and feeds SMR for methane conversion into syngas via Equations 3–5 (Zanetti et al., 2025).

$$C_n H_m + nH_2 O \to nCO + [n + (m/2)]H_2 \quad ; \Delta H > 0$$
 (1)

$$CO + 3H_2 \leftrightarrow CH_4 + H_2O$$
 ; $\Delta H < 0$ (2)

$$CO + H_2O \leftrightarrow CO_2 + H_2$$
 ; $\Delta H < 0$ (3)

$$CH_4 + H_2O \leftrightarrow CO + 3H_2$$
 ; $\Delta H > 0$ (4)

$$CH_4 + 2H_2O \leftrightarrow CO_2 + 4H_2$$
; $\Delta H > 0$ (5)

Several heat integrations are implemented because process water needs to be pre-heated before it is fed to the HRSG for steam generation, and hot syngas must cool; that is, (i) before and after the high-temperature WGS reactor (WGS-HT); (ii) after the low-temperature WGS reactor (WGS-LT). Hot syngas ($T > 700~^{\circ}$ C) from the SMR cools to $340~^{\circ}$ C and feeds the WGS-HT, whose outletshifted syngas cools to $242~^{\circ}$ C and feeds the WGS-LT. WGS-LT-shifted syngas cools to $151~^{\circ}$ C and cools again to $40~^{\circ}$ C with CW, and a flash separates process water from shifted syngas. Approximately 26.4% of the shifted syngas recycles to pre-reforming, while the rest is decarbonized with aqueous MDEA-PZ. The flash process water mixes with process water from the RANK and recycles to the HRSG for SMR steam generation.

The kinetic models for pre-reforming, SMR, and WGS reactors are based on Zanetti et al. (2025) and are discussed in Supplementary Appendix A. These models were installed in HYSYS for HU simulation and design.

Shifted syngas ($40.0~^{\circ}$ C, $19.5~^{\circ}$ bar, $\approx 65\%~^{\circ}$ mol H_2 , $\approx 30\%~^{\circ}$ mol CO_2) is fed at the bottom of the 10-stage absorber with aqueous MDEA-PZ ($MDEA = 40\%~^{\circ}$ w/w, $PZ = 5\%~^{\circ}$ w/w). Decarbonized H_2 is obtained at the top and sent to the GT. CO_2 -rich aqueous MDEA-PZ is preheated, then depressurized, and feeds Stage #5 of the low-pressure stripper for solvent regeneration. The stripper operates with a total reflux condenser at $40~^{\circ}$ C, and its reboiler is heated with LPS. Lean solvent leaves the reboiler at $\approx 100~^{\circ}$ C, is cooled first with rich solvent, then with CW, receives make-up, and is finally pumped back to the absorber (da Cunha et al., 2023).

2.1.7 Base case

The base case is a typical NGCC power plant without carbon capture, depicted in Figure 4 as a block diagram using the subsystems GT (Figure 1a) and RANK (Figure 3b). Table 2 shows the base case operational assumptions.

2.1.8 NGCC with anti-carbon packages

Figure 5 depicts block diagrams for the three NGCCs with anti-carbon packages studied in this work. The blocks used in Figure 5 correspond to the subsystems defined in Figures 1–3, namely, (i) post-combustion NGCC using blocks GT, RANK, PCC-MEA, and CC-TEG (Figure 5a); (ii) oxy-combustion NGCC with blocks ASU, GT, RANK, and CC-TEG (Figure 5b); and (iii) precombustion NGCC using blocks HU, GT, RANK, PCC-MEA, and CC-TEG (Figure 5c). Table 2 shows the corresponding operational assumptions.

2.1.9 Operational and design assumptions

Table 2 depicts the simulation and design assumptions used in this work.

2.2 Preliminary economic analysis of the process

Preliminary economic analysis of the process uses the method of Turton et al. (2018). Turton et al. (2018) emphasize that this analysis is strictly preliminary in the sense that error magnitudes can be as high as $\pm 40\%$. Nevertheless, it can be very useful for comparing process alternatives as well as for eliminating very bad ones from

TABLE 2 Simulation and design assumptions.

ltem	Topic	Assumption
A1	Simulation software	Aspen HYSYS 14
A2	Thermodynamic modeling	ASU, HU, GT: HYSYS Peng–Robinson equation of state Pre-reforming, SMR, WGS reactors: Supplementary Appendix A Aqueous MEA, aqueous MDEA-PZ: HYSYS acid–gas chemical solvents package CC-TEG: HYSYS Twu–Sim–Tassone equation of state RANK: HYSYS ASME table; LPS, HPS, ChW, CW: HYSYS ASME table
B1	NG	$6.6 \text{MMSm}^3/\text{d} (274,541.008 \text{kmol/d}), P = 25 \text{bar}, T = 40 ^\circ \text{C}, \text{dry basis: CH}_4 = 50\% \text{mol}, \text{CO}_2 = 45\% \text{mol}, \text{C}_2\text{H}_6 = 3\% \text{mol}, \text{C}_3\text{H}_8 = 1\% \text{mol}, \text{C}_4\text{H}_{10} = \text{iC}_4\text{H}_{10} = 0.5\% \text{mol}; \text{water saturated; CO}_2^{ \text{Equiv}} - \text{inlet} = 295,629.599 \text{kmol/d}$
B2	Air	$25 ^\circ \!\! \text{C, } 1.013 \text{bar, dry basis: N}_2 = 78.08\% \text{mol, O}_2 = 20.95\% \text{mol, Ar} = 0.93\% \text{mol, CO}_2 = 0.04\% \text{mol; humidity}^{\text{Relative}} = 60\% \text{mol, CO}_2 = 0.04\% \text{mol, CO}_2 = $
В3	CW	T ^{Inlet} = 35 °C; P ^{Inlet} = 4 bar; T ^{Outlet} = 50 °C; P ^{Outlet} = 3.5 bar
B4	CW tower	$\Delta P^{Fan} = 0.013$ bar; blowdown = evaporation; $T^{Make-up} = 25$ °C
B5	ChW	$T^{\text{Inlet}} = 15 ^{\circ}\text{C}; P^{\text{Inlet}} = 4 \text{bar}; T^{\text{Outlet}} = 35 ^{\circ}\text{C}; P^{\text{Outlet}} = 3.5 \text{bar}$
C1	Machines	Adiabatic efficiencies: $\eta^{Pump} = \eta^{Fan} = \eta^{Blower} = \eta^{Compressor} = \eta^{Expander} = 75\%$ Compressors: compression ratio $Stage \le 3$
C2	DCC with CW	3-Stage; $\Delta P^{Stage} = 1$ kPa; flue gas: $T^{Outlet} \le 40$ ℃
С3	DCC with ChW	3-Stage; $\Delta P^{Stage} = 1$ kPa; air: $T^{Outlet} \le 20$ ℃
C4	Capture efficiency	$\eta^{AqueousMEA} \geq 90\%; \eta^{AqueousMDEA-PZ} \geq 90\%$
D1	LPS (saturated)	P=4.76 bar, $T=150$ °C; oxy-combustion: $P=23.2$ bar, $T=220$ °C
D2	HPS (superheated)	$P = 35 \text{ bar}, T = 625 ^{\circ}\text{C}$
D3	Thermal approaches	$\Delta T^{Gas\text{-}Gas} \geq 25 \text{ °C}; \ \Delta T^{Gas\text{-}Liq} \geq 5 \text{ °C}; \ \Delta T^{Liq\text{-}Liq} \geq 5 \text{ °C}; \ \Delta T^{Cold\ box} \geq 1 \text{ °C}$
E1	RANK ^a	Steam turbine: $P^{\text{Discharge}} = 0.18 \text{ bar}$; quality $P^{\text{Discharge}} \ge 99\%$
E2	CC-TEG	Absorber: 20-Stage; $P^{\text{Bottom}} \approx 64.8 \text{ bar}$; TEG = 95% w/w; H ₂ O = 5% w/w Capture ratio = 6 kg ^{TEG} /kg ^{H2O} ; $\Delta P^{\text{Stage}} = 1 \text{ kPa}$ Stripper: 10-Stage; $P^{\text{Cond}} \approx 1.2 \text{ bar}$; heat ratio = 60 kJ/mol ^{H2O} ; $T^{\text{Reboiler}} < 140 ^{\circ}\text{C}$
E3	CO ₂ -to-EOR	$p^{\text{Dispatch}} = 250 \text{ bar}; H_2O \le 50 \text{ ppm-mol}$
E4	PCC-MEA	Absorber: 40-Stage; capture ratio = 12 kg ^{Solvent} /kg ^{CO} ₂ ; MEA = 30% w/w; H ₂ O = 70% w/w; P^{Top} = 1.013 bar; ΔP^{Stage} = 1 kPa Stripper: 20-Stage; heat ratio ≤ 200 kJ/mol ^{CO} ₂ ; T^{Cond} = 40 °C; P^{Cond} ≈ 1 atm
E5	ASU	HPC: $P^{\text{Top}}=4.3 \text{ bar}$; LPC: $P^{\text{Top}}=1.07 \text{ bar}$; $O_2 \geq 96\% \text{ mol}$; $N_2 \geq 98.5\% \text{ mol}$
E6	ASU TSA	$\begin{aligned} & \text{Loading}^{\text{Al2O3}} = 0.1 \text{ kg}^{\text{H2O}}/\text{kg}^{\text{Al2O3}}; \text{loading}^{\text{MS-13X}} = 0.185 \text{ kg}^{\text{CO}}_{2}/\text{kg}^{\text{MS-13X}} \\ & \text{Bed regeneration: } T^{\text{Al2O3}} \geq 200 \text{ °C}; T^{\text{MS-13X}} \geq 120 \text{ °C} \end{aligned}$
E7	Gas turbine ^a	Siemens SGT6-8000HL heavy-duty (310 MW); H ₂ -GT: air Excess = 177%
E8	HU SMR-WGS	$Burner^{SMR}; NG^{Fraction} = 55\%; air^{Excess} = 50\%; heat^{Loss} = 10\%$ $Pre-reforming; NG^{Fraction} = 45\%; P^{NG} = 28.3 \ bar; T^{NG} = 390 \ ^{\circ}C; T^{Steam} = 500 \ ^{\circ}C$ $SMR^{Inlet}; P^{Syngas} = 27.8 \ bar; T^{Syngas} = 540 \ ^{\circ}C; WGS-HT^{Inlet}; P^{Syngas} = 24.5 \ bar; T^{Syngas} = 242 \ ^{\circ}C$ $Shifted \ syngas; P = 19.50 \ bar; T = 40 \ ^{\circ}C; recycle-to-SMR = 26.41\%$
E9	HU shifted syngas decarbonation	Absorber: 10-Stage, capture ratio = $15 \text{kg}^{\text{Solvent}}/\text{kg}^{\text{CO}}_2$ Solvent: MDEA = 45% w/w, PZ = 5% w/w, H ₂ O = 50% w/w Stripper: 10-Stage; T ^{Cond} = 40 °C; T ^{Reboiler} \leq 110 °C

 $^{^{\}mathrm{a}}$ Steam turbine capacity reduction can be handled via electro-mechanical devices; NG gas turbine is adjustable to burn H_2 .

TABLE 3 $\,$ Economic assumptions for the preliminary economic analysis of the process.

Economic parameter			
CEPCI	816 (reference date: December 2022) (Rodrigues et al., 2024)		
Project horizon (includes construction)	25 years		
Construction	2 years (Rodrigues et al., 2024)		
Investment implementation for construction	40%/60% FCI (Rodrigues et al., 2024)		
Plant operation	8,400 h/y (Rodrigues et al., 2024)		
Annual depreciation (MMUSD/y)	10% FCI (Turton et al., 2018)		
Income tax rate (ITR)	34% (Turton et al., 2018)		
Annual interest rate (i)	6% (Rodrigues et al., 2024)		
Commodity prices at the re	eference date		
NG	2.53 USD/MMBTU		
Demineralized water	0.5 USD/m³ (Rodrigues et al., 2024)		
MEA	2000 USD/t (Milão et al., 2023)		
MDEA	2100 USD/t (PW Consulting Chemical & EnergyResearch Center, 2024)		
PZ	101 USD/t (Researz, 2024)		
TEG	1,216 USD/t (Chemanalyst, 2024)		
Activated alumina	1,357 USD/t		
Zeolite-13X	2,060 USD/t		
Electricity	84.5 USD/MWh (Rodrigues et al., 2024)		
Oil	70 USD/bbl ^{Oil} (Rodrigues et al., 2024)		
Special costs and revenues	;		
Unitary revenue from CO ₂ to EOR	1 bbl ^{Oil} /t ^{CO} ₂ (McCoy, 2008)		
Unitary cost of CO ₂ transportation via large-capacity horizontal pipeline (after inflationary correction of typical value from McCoy (2008))	3 USD/(t ^{CO2} .100 km)		
Horizontal pipeline length for ${\rm CO}_2$ transport	275 km		

further consideration. Table 3 shows economic assumptions for the preliminary economic analysis of the process.

After defining the reference date of the analysis, the chemical engineering plant cost index (CEPCI) must be found in the

literature for correcting equipment prices to the reference date, taking into account inflationary market factors. The bare module cost $C_{BM}(MMUSD)$ refers to equipment cost at process conditions and is estimated from correlations by Turton et al. (2018) using the bare module cost at the reference conditions C^{o}_{BM} (MMUSD), with correlations from Turton et al. (2018). Bare module costs out of the correlation range can be estimated via Equation 6 using limit values, where CF represents a characteristic capacity factor (e.g., power (kW) for machines, area (m²) for heat exchangers, etc.). The grassroots equipment cost C_{GR} (MMUSD) in Equation 7 refers to installed equipment at process conditions in a new plant. The fixed capital investment FCI (MMUSD) of a new plant is given by Equation 8, where NEQ is the number of equipment items. The number of operators (N_{OL}) working 4.5 shifts per week is estimated by Equation 9 with NEQ. The cost of labor COL (MMUSD/y) is calculated using Equation 10 with the operator cost Op_{COST} (MMUSD/y). The cost of manufacturing COM (MMUSD/y) is estimated by Equation 11 using FCI (MMUSD), the cost of utilities CUT (MMUSD/y), the cost of raw materials CRM (MMUSD/y), and the cost of CO₂ transportation CCT (MMUSD/y). Gross annual profit GAP (MMUSD/y), or profit before taxes, is calculated using Equation 12 with COM (MMUSD/y) and revenues REV (MMUSD/y). Annual profit AP (MMUSD/y) is calculated with Equation 13 using GAP (MMUSD/y), depreciation DEPR (MMUSD/y), and the income tax rate ITR(%). Finally, the net present value NPV (MMUSD) is obtained via Equation 14 using the annual interest rate i (%), the project horizon NH (years), and the annual profit of the k^{th} year AP_k (MMUSD/y). Equation 14 assumes the investment (FCI) is implemented in the first two project years at 40% and 60%, respectively, and production only exists after the second year ($AP_k = 0$ for k < 3). In addition, investment expenses are made at the beginning of the year, while revenues are consolidated at the end of the year.

$$\frac{C_{BM}}{C_{BM}^{\lim}} = \left(\frac{CF}{CF^{\lim}}\right)^{0.6},\tag{6}$$

$$C_{GR} = 1.18^* C_{BM} + 0.50^* C_{BM}^o, (7)$$

$$FCI = \sum_{j}^{NEQ} C_{GR}(j), \tag{8}$$

$$N_{OL} = 4.5(6.29 + 0.23NEQ)^{0.5}, (9)$$

$$COL = N_{OL}Op_{COST}, (10)$$

$$COM = 0.18FCI + 2.73COL + 1.23(CUT + CRM + CCT),$$
 (11)

$$GAP = REV - COM, (12)$$

$$AP = \begin{cases} GAP - (GAP - DEPR) \left(\frac{ITR}{100}\right) (GAP > DEPR), \\ GAP & (GAP \le DEPR), \end{cases}$$
(13)

$$NPV = -\left(0.4 + 0.6q^{-1}\right)FCI + \left(\sum_{k=0}^{NH} AP_k^* q^{-k}\right) \; ; \quad q = \left[1 + \left(\frac{i}{100}\right)\right]. \tag{14}$$

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Bare module costs of special equipment and blocks that do not exist in the correlations of Turton et al. (2018) were found in the literature as follows: (i) the C_{BM} of a complete CW tower system (C_{BM}^{CWS}) is calculated with Equation 15 where the reference value $(C_{BM}^{CWS,ref})$ is 1.2 MMUSD for a reference CW flow rate of 1,000 kg/s (CF^{ref}) (Woods, 2007); (ii) C_{BM} (MMUSD) of a GT block $\left(C_{BM}^{GT}\right)$ is calculated by Equation 16 where W^{GT} (kW) is the respective power output (Wang et al., 2023); and (iii) C_{BM} (MMUSD) of the HRSG $\left(C_{BM}^{HRSG}\right)$ is calculated with Equation 17 where W^{RANK} (kW) is the power output of the respective Rankine cycle (Middelhoff et al., 2022).

$$C_{BM}^{CWS} = 1.22 \times \left[C_{BM}^{CWS,ref} \left(\frac{CF}{CF^{ref}} \right)^{0.64} \right],$$
 (15)

$$C_{BM}^{GT} = W^{GT} \left(0.108^* W^{GT-0.53} + 4.9^* 10^{-5} \right) \frac{CEPCI}{708.8},$$
 (16)

$$C_{BM}^{HRSG} = 0.1352175^* W^{RANK*} \frac{CEPCI}{567.3}.$$
 (17)

2.3 Marginal abatement cost for CCHU/CCS technologies

The marginal abatement cost or MAC (USD/t^{CO2}) is the increment of the COM (MMUSD/y) (Equation 11) of a process to capture an additional CO2 ton per year (Huang et al., 2016). The economic-environmental comparison of power production technologies with CCU can be implemented through MAC (Eory et al., 2018).

As a marginal cost, MAC is the derivative shown in Equation 18. This equation can be approximated via finite differences in Equation 19. Because extra revenues normally arise with CCU, MAC is generalized in Equation 20 using the GAP (MMUSD/y) from Equation 12. Consequently, negative MAC values can occur and mean that sufficient CCU gains appeared, afforded CO₂ abatement, and even generated profits.

Now, because money value is time-dependent and because a process normally experience changes in REV, COM, and CCU^{Capacity} along the project horizon (NH or N, years), it is better to redefine MAC for the entire project campaign in terms of the accumulated discounted GAP, or AGAP (MMUSD), using the interest rate i (%) over N years, as shown in Equations 21–23.

$$MAC(USD/t^{CO2}) = \frac{dCOM^{Process+CCU}(USD/y)}{dCCU^{Capacity}(t^{CO2}/y)},$$
(18)
$$MAC(USD/t^{CO2}) \cong \frac{COM^{Process+CCU}(USD/y) - COM^{Process}(USD/y)}{CCU^{Capacity}(t^{CO2}/y)},$$
(19)

$$CCU^{Capacity}(t^{CO2}/y)$$
 $CCU^{Capacity}(t^{CO2}/y)$
 $CCU^{Capacity}(t^{CO2}/y)$
 $CCU^{Capacity}(t^{CO2}/y)$
 $CCU^{Capacity}(t^{CO2}/y)$
 $CCU^{Capacity}(t^{CO2}/y)$

$$MAC(USD/t^{CO2}) = \frac{GAP^{Process}(USD/y) - GAP^{Process+CCU}(USD/y)}{CCU^{Capacity}(t^{CO2}/y)},$$
(20)

$$MAC(USD/t^{CO2}) = \frac{AGAP^{Process}(USD) - AGAP^{Process+CCU}(USD)}{CCU^{Total}(t^{CO2})},$$
(21)

$$AGAP^{Process}(USD) = \sum_{k=0}^{N} \frac{GAP_{k}^{Process}}{(1+i/100)^{k}}, AGAP^{Process+CCU}(USD)$$
$$= \sum_{k=0}^{N} \frac{GAP_{k}^{Process+CCU}}{(1+i/100)^{k}}, \tag{22}$$

$$CCU^{Total}(t^{CO2}) = \sum_{k=0}^{N} CCU_k^{Capacity}(t^{CO2}/y). \tag{23}$$

3 Results and discussion

The technical, environmental, economic, and MAC results are discussed and compared with literature analogs.

3.1 Simulation results

The four studied NGCC—base case NGCC, post-combustion NGCC, oxy-combustion NGCC, and pre-combustion NGCC—were simulated at a steady state and solved in HYSYS 14. Certain critical and representative operations of post-combustion NGCC, oxycombustion NGCC, and pre-combustion NGCC were selected in Figure 6 for a graphic demonstration of the correct implementation of flowsheets (Figures 1-5) and technical assumptions (Table 2). In this regard, Figure 6a depicts the thermal composite curves in the HRSG of subsystem RANK in the post-combustion NGCC; Figure 6b depicts the thermal composite curves in the heat exchanger for thermal integration of LPC/HPC distillation columns of subsystem ASU in the oxy-combustion NGCC; Figure 6c depicts H₂, CO, CO₂, H₂O, and CH₄ mole fraction profiles along the steam reforming (SMR) reactor of subsystem HU in the pre-combustion NGCC; and Figure 6d depicts temperature profiles of the reforming stream and hot flue gas stream along the same SMR reactor of HU in the pre-combustion NGCC. All variables behave as expected vis-à-vis the technical assumptions (Table 2).

3.2 Technical and environmental results

Table 4 presents the technical and environmental results of NGCC configurations. An environmental impact is water consumption (process water make-up plus CW make-up). It is seen that pre-combustion NGCC has the largest CW circulation due to high CW consumption in: (i) shifted syngas decarbonization with aqueous MDEA-PZ, (ii) PCC-MEA decarbonization of flue gas from SMR burners, (iii) cooling after WGS-LT, and (iv) CO2 multi-stage intercooled compression. Oxy-combustion NGCC CW consumption comes second due to the high cooling demand of its massive intercooled air compression train and its intercooled CO2 compression train. Meanwhile, post-combustion NGCC demands less cooling in its intercooled CO2 compression train and PCC-MEA condensers and coolers. The pre-combustion NGCC, with its highest CW consumption, naturally entails the highest CW make-up, that is, the worst water-intensity environmental result. The pre-combustion NGCC also comes second as a CO2 emitter among decarbonized NGCCs, while the post-combustion NGCC is the greatest emitter

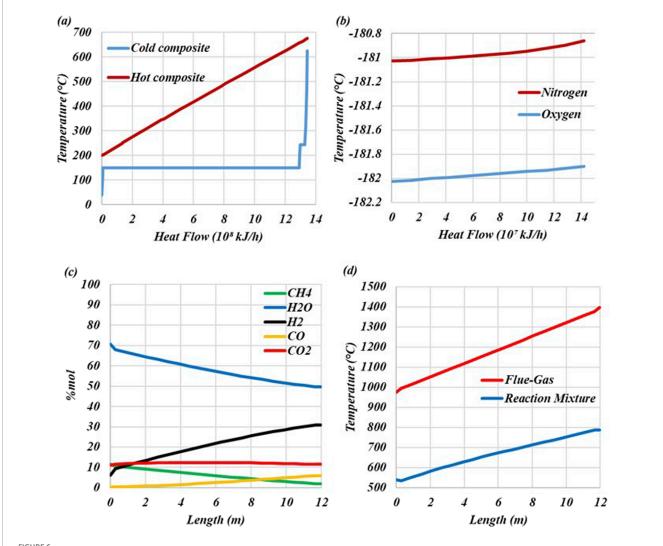


FIGURE 6
Profiles in the critical operations of decarbonized NGCCs: (a) thermal composite curves in the HRSG of the RANK (post-combustion NGCC); (b) thermal composite curves along the LPC/HPC integration exchanger of the ASU (oxy-combustion NGCC); (c) composition profiles vs. axial position along the SMR reactor of the HU (pre-combustion NGCC); (d) temperature profiles of the reforming stream and flue gas stream along the SMR reactor of the HU (pre-combustion NGCC).

among the decarbonized NGCCs. The oxy-combustion NGCC has the lowest ${\rm CO_2}$ emission (lowest carbon intensity).

The oxy-combustion NGCC also presents the lowest LPS consumption, despite its highest power consumption due to its massive air compression train in the ASU. In contrast, both the post-combustion NGCC and the pre-combustion NGCC require approximately 50% less power consumption. However, high LPS consumption is required for solvent regeneration in the $\rm CO_2$ strippers in the decarbonation steps, though a little less is needed in the pre-combustion NGCC, as aqueous MDEA-PZ demands a lower heat ratio for regeneration relative to aqueous MEA.

An evident advantage of the oxy-combustion NGCC is its near-zero CO_2 emission, as NG combustion is performed with pure oxygen, creating a $\mathrm{CO}_2+\mathrm{H}_2\mathrm{O}$ flue gas from which CO_2 is easily captured with $\approx 100\%$ efficiency. In contrast, post-combustion NGCC and pre-combustion NGCC require flue gas decarbonization in PCC-MEA. wherein CO_2 capture efficiency rarely exceeds

90%, before atmospheric release. Flue gas decarbonization in pre-combustion NGCC is even more difficult than in the post-combustion NGCC counterpart because the flue gas flow rate in the latter is lower while the CO_2 content is higher, due to utilization of EGR, which reduces air intake to a stoichiometric flow rate, while the PCC-MEA in the pre-combustion NGCC does not operate with EGR, entailing a lower CO_2 content in the flue gas simultaneous with the high flow rate.

Not surprisingly, the largest CO_2 -to-EOR comes from the oxycombustion NGCC, a consequence of its highest CO_2 capture efficiency $\approx 100\%$. Meanwhile, the post-combustion NGCC has the lowest CO_2 -to-EOR, as the PCC-MEA operates with a CO_2 capture efficiency $\approx 90\%$. Shifted syngas decarbonization with aqueous MDEA-PZ in the pre-combustion NGCC has a capture efficiency $\approx 95\%$.

The post-combustion NGCC has the highest gas turbine power, as the oxy-combustion NGCC burns NG with $\rm O_2$, generating $\rm CO_2$

TABLE 4 Technical and environmental results of the NGCC configurations.

	Base case NGCC (no CCU)	Post-combustion NGCC	Oxy-combustion NGCC	Pre-combustion NGCC			
Start-up inventories							
Water (m³)	24,922.80	43,747.11	35,207.89	37,131.92			
Aqueous MEA 30% w/w (t)	0.00	6,861.28	0.00	5,679.56			
Aqueous MDEA-PZ 45%/5% w/w (t)	0.00	0.00	0.00	1,118.57			
TEG 99% w/w (t)	0.00	98.74	130.90	142.12			
Raw materials							
NG (MMSm³/d)	6.60	6.60	6.60	6.60			
Fossil carbon intake (kmol/h)	12,517.83	12,517.83	12,517.83	12,517.83			
Process water make-up (t/h)	0.00	0.00	0.00	0.00			
CW make-up (t/h)	2,016.76	2,470.27	2,772.55	3,137.70			
Aqueous MEA make-up (t/h)	0.00	$H_2O = 133.86$ MEA = 0.0032	0.00	$H_2O = 112.22$ MEA = 0.52			
Aqueous MDEA-PZ make-up (t/h)	0.00	0.00	0.00	$\begin{aligned} \mathbf{H}_2\mathbf{O} &= 6.70\\ \mathbf{MDEA} &= \mathbf{PZ} = 0 \end{aligned}$			
TEG 99% w/w make-up (kg/h)	0.00	5.16	4.81	3.44			
Utilities							
CW (t/h)	48,460.99	55,464.52	69,230.22	74,255.40			
LPS (t/h)	0.00	1,031.42	25.54	961.04			
Outlets							
CO ₂ emission (t/h)	550.785	60.402	10.555	48.380			
Treated flue gas ($CO_2 \le$ 1.53% mol) (t/h)	0.00	1,617.57	0.00	4,108.13			
CO ₂ -to-EOR (t/h)	0.00	490.38	540.23	501.45			
Power and capture penalty							
Gas turbines (MW)	639.25	616.56	579.61	344.35			
Steam turbine (MW)	215.76	43.46	207.19	92.81			
Gross power (MW)	855.01	660.02	786.82	437.16			
Power consumption (MW)	15.25	101.75	241.91	126.99			
Net power (MW)	839.76	558.27	544.91	310.17			
Electricity capture penalty (MW)		281.49	294.85	529.59			
Capture penalty (MWh/t ^{CO2})		0.574	0.546	1.056			
Power ^{Exported-25y} (MWh)	162,241,632	107,857,764	105,276,612	59,924,844			

+ $\rm H_2O$, without nitrogen. Because $\rm CO_2$ and $\rm H_2O$ have greater heat capacities than $\rm N_2$, the oxy-combustion flame temperature and flue gas flow rate are both lower, entailing a lower Brayton cycle thermodynamic yield. The pre-combustion NGCC has the lowest gas turbine power because 55% of NG is diverted to heat the SMR reactors, while only 45% of NG is converted into $\rm H_2$ for power production.

The oxy-combustion NGCC has the largest team-turbine power due to its lowest LPS demand, entailing that its Rankine cycle (RANK) operates with the highest HPS flow rate. As both the pre-combustion NGCC and the post-combustion NGCC have large LPS demands, the oxy-combustion NGCC combined-cycle power (gross power) is the highest. However, the oxy-combustion NGCC also has the highest power consumption (for ASU air compression), entailing that the post-combustion NGCC has the highest net power. The pre-combustion NGCC power performance is the worst among the CCUS-NGCCs. These efficiency differences are also perceived in terms of electricity capture penalty (MW) and capture penalty (MWh/tCO2), where the former represents net power loss after decarbonization, and the latter is the ratio between the electricity capture penalty (MW) and the CO2-to-EOR (t^{CO2}/h) ratio. Table 4 depicts capture penalties for the postcombustion NGCC, oxy-combustion NGCC, and pre-combustion NGCC, which are, respectively, 0.574 MWh/t^{CO2}, 0.546 MWh/t^{CO2}, and $1.056~MWh/t^{\rm CO2}$. The lowest capture penalty of the oxycombustion NGCC reflects its greatest power generation and greatest CO2-to-EOR among CCUS-NGCCs.

3.3 Economic and MAC results

Table 5 presents the economic and *MAC* results. All NGCCs with anti-carbon packages have remarkably larger *FCI* and lower electricity revenues than the base case, so that the environmental benefit should be large enough to compensate for greater investments, higher costs, and revenue losses.

As expected, the post-combustion NGCC requires the lowest *FCI*. Although the high investment of pre-combustion NGCC, the highest *FCI* comes from the oxy-combustion NGCC, as cryogenic ASU equipment, which is made of titanium and aluminum alloys, is expensive. The oxy-combustion NGCC presents the highest *COM* due to the highest *FCI*, in accordance with Equation 11, as the other contributing costs of *COM* have similar magnitudes.

CCT and CO_2 monetization revenues depend on the value of CO_2 -to-EOR, which attains the highest value in the oxycombustion NGCC, while the post-combustion NGCC has the lowest. In addition, the post-combustion NGCC presents the highest electricity revenues, which are remarkably lower ($\approx 45\%$ lower) in the pre-combustion NGCC. The largest total revenues come from the oxy-combustion NGCC, thanks to its highest CO_2 capture, while the pre-combustion NGCC has the lowest total revenues.

Considering CO_2 monetization, the three NGCCs with anticarbon packages are profitable with positive NPV^{25y} and acceptable payback periods. Among the NGCCs with anti-carbon packages, the post-combustion NGCC attains the highest NPV^{25y} , while the pre-combustion NGCC attains the lowest.

The MACs of the CCU alternatives were calculated for comparisons using the $AGAP^{25y}$, the 25 years accumulated

discounted GAP, for the base case and all NGCCs with anticarbon packages, and CCU25y, which is the total CO2 captured during project lifetime. The pre-combustion NGCC attained the highest MAC; that is, it has the worst economic performance among all NGCCs with anti-carbon packages, so that carbon capture implementation on the base case NGCC attained the highest unitary cost or the lowest AGAP^{25y}. The post-combustion NGCC attained the lowest MAC because its $AGAP^{25y}$ showed a lesser reduction from the base case AGAP^{25y} compared to the oxy-combustion NGCC counterparts, whose large decrease of AGAP^{25y} is a symptom of their high COM (and high FCI). Figure 7 presents the NPV vs. project lifetime for the base case NGCC and all NGCCs with anti-carbon packages. The payback times correspond to the respective times for zero NPV. As expected, the pre-combustion NGCC developed the lowest NPV^{25y} in spite of the highest FCI of the oxy-combustion NGCC, which could recover a higher NPV^{25y} due to its highest CO₂-to-EOR revenues.

The performance ratios (Table 5) are useful to measure performance disconnected from scale. In this sense, the base case NGCC evidently has the best Power Exported-25y/FCI because it simultaneously has the lowest investment and the highest power exported, while the three decarbonized NGCCs must pay for their capture penalties. The ratio Power Exported-25y/NPV has an inverted sense because it ranks NGCCs from the highest NPV to the lowest (i.e., the one with the worst economic return) for similar power exports. It is not surprising to see the leadership of the precombustion NGCC according to this indicator. The fact that the base case is located ahead of the post-combustion NGCC, which is theoretically the best NGCC according to this metric, has to do with its much greater power export than all three CCU NGCCs. However, the true performance discriminator among these performance ratios undoubtedly is Power Exported/CO₂ Emitted (MWh/t^{CO2}), which reveals how much social benefit (electricity) per unit of environmental harm (carbon emission) is being provided. It is amazing to see how the oxy-combustion NGCC is more efficient than the other NGCCs according to this metric. Its Power Exported / CO2 Emitted is, respectively, approximately 5.5× and 8× the counterparts of post-combustion NGCC and pre-combustion NGCC (and 34x the base case NGCC counterpart). Consequently, the oxy-combustion NGCC clearly has the greatest sustainability among the NGCCs with anti-carbon packages despite its second place in water intensity.

3.4 Comparisons with literature

The results obtained here were compared with technoenvironmental-economic data from decarbonized NGCC available in the literature. In the cases where *MAC* evaluations are missing, the respective *MAC* values were indirectly calculated using other kinds of information available in the same source.

Isogai and Nakagaki (2024) studied a post-combustion NGCC, while Manzolini et al. (2013) assessed both post-combustion and pre-combustion NGCCs. In both studies, an NGCC without carbon capture was presented and was used as the base case for calculating *MACs*. Table 6 presents these results. In addition to *MAC*, *FCI* and *NPV* were also compared. For the sake of normalizing the data and to eliminate the influence of scale differences, some performance ratios were also compared. Isogai and Nakagaki (2024) incorporated

TABLE 5 Economic and MAC results for NGCC configurations.

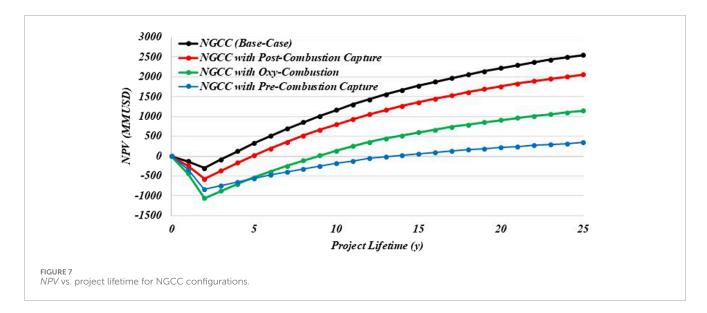
	Base case NGCC	Post-combustion NGCC	Oxy-combustion NGCC	Pre-combustion NGCC				
FCI (MMUSD)	318.25	596.89	1,095.19	874.72				
ΔFCΙ		278.64	776.94	556.47				
Costs (MMUSD/y)								
COL	0.78	12.57	12.66	1.37				
CUT	0.00	0.00	0.00	0.00				
CRM	127.51	125.74	126.61	141.55				
CCT	0.00	33.98	37.44	34.75				
COM	216.24	338.23	433.47	378.04				
Revenues (MMUSD/y)								
Electricity	602.41	396.41	386.78	220.16				
CO ₂ -to-EOR	0.00	288.35	317.66	294.85				
Revenues	602.41	684.76	704.44	515.01				
Profits and depreciation (N	MMUSD/y)							
GAP	386.18	346.53	270.97	136.97				
DEPR (3rd-12th year)	31.82	59.69	109.52	87.47				
DEPR (13th-25th year)	0.00	0.00	0.00	0.00				
Income tax (3rd-12th year)	120.48	97.53	54.89	16.83				
Income tax (13th–25th year)	131.30	117.82	92.13	46.57				
AP (3rd–12th year)	265.70	249.00	216.08	120.14				
AP (13th-25th year)	254.88	228.71	178.84	90.40				
Net present value								
NPV ^{25y} (MMUSD)	2,554.33	2,060.68	1,144.21	339.66				
Payback time (y)	≈4	≈5	≈9	≈14				
MAC								
AGAP ^{25y} (MMUSD)	4,228.63	3,794.49	2,967.11	1,499.79				
CCU ^{25y} (MMt ^{CO2})	0.00	94.73	104.37	96.85				
MAC (USD/t ^{CO2})		4.58	12.09	28.17				
Performance ratios								
Power ^{Exported-25y} /NPV (MWh/MMUSD)	63,516.32	52,340.86	92,008.12	176,425.97				
Power ^{Exported-25y} /FCI (MWh/MMUSD)	509,793.06	180,699.57	96,126.34	68,507.46				

(Continued on the following page)

TABLE 5 (Continued) Economic and MAC results for NGCC configurations.

	Base case NGCC	Post-combustion NGCC	Oxy-combustion NGCC	Pre-combustion NGCC
Power ^{Exported} /CO ₂ Emitted (MWh/t ^{CO2})	1.5247	9.2426	51.6258	6.4111
$FCI^{Capture\ penalty} = \Delta FCI/CCU^{25y}$ (USD/t^{CO2})		2.941	7.444	5.746

A bold value represents the total value or the most important value of a given category of items.



a carbon tax in their analysis, which imposes an additional cost on the base case NGCC. As a result, the decarbonized NGCC yields a higher gross annual profit, which is reflected in the calculated *MAC*, yielding a negative value. Their work considers only a retrofit scenario, focusing exclusively on the investment related to the capture facility. However, for consistency and comparability, the present comparison took into consideration the full *FCI* of their process in the analysis.

Manzolini et al. (2013) did not specify the electricity price, as their objective was to determine the break-even electricity price; that is, the price at which electricity must be sold to achieve zero NPV over the project lifetime. To align their results with the present methodology, the same electricity price defined in Table 3 was applied. A consequence of this choice was that their decarbonized NGCC plants did not achieve the break-even point; that is, negative NPV resulted. Another observation regarding Manzolini et al. (2013) is that their pre-combustion NGCC was reported to achieve a power yield—the so-called LHV efficiency—higher than their post-combustion NGCC counterpart, which is a somewhat atypical result. Consequently, the MAC values do not vary significantly between their NGCC configurations, yet they remain relatively high, particularly for the post-combustion NGCC, due to the absence of carbon utilization revenues. Another aspect of Manzolini et al. (2013) is that they did not inform the CO₂ emission associated with their decarbonized NGCCs. This information is critical for the complete evaluation of the sustainability performance of decarbonized plants.

Table 6 confirms that the oxy-combustion NGCC achieves the highest power exported per unit of CO_2 emitted, reflecting its near-zero emission profile. This comes at the cost of higher FCI and $FCI^{Capture\ penalty}$ (Table 5), which limit its economic attractiveness despite the environmental benefit. However, it is important to register that oxy-combustion NGCCs are the most sustainable fossil power production systems as they maximize social benefit (electricity) per unit of climatic harm (fossil carbon emission).

Points on the comparison of performance ratios (Table 6) are raised in the following paragraphs.

Power Exported / FCI (MWh/MMUSD) is a useful ratio related to the social efficiency of the decarbonized NGCC, as it unveils how much social benefit appears per USD invested. According to this ratio, the best sustainable plant is the post-combustion NGCC (this work), characterized by low FCI, which leads to a pragmatic conclusion: "with low money, prioritize post-combustion NGCC." This aphorism is corroborated by the post-combustion NGCC of Manzolini et al. (2013), which comes second. The counterpart of Isogai and Nakagaki (2024) did not confirm the rule due to its anomalously low power exported, while all the pre-combustion NGCCs and the oxy-combustion NGCC are handicapped by high FCI.

NPV/Power^{Exported} (USD/MWh) reports plant profitability per unit of social benefit. It is a somewhat egoistic ratio that privileges the benefit of the plant owner in the sense that high values derive from high NPV and/or low social benefit. Here, the post-combustion NGCC (this work) is again the best option thanks to its highest NPV.

TABLE 6 Comparison of decarbonized NGCCs with literature.

	Post-combustion NGCC			Pre-combustion NGCC		Oxy- combustion NGCC
References	Present work	Manzolini et al. (2013)	Isogai and Nakagaki (2024)	Present work	Manzolini et al. (2013)	Present work
FCI (MMUSD)	596.89	860.63	776.20	874.72	1,215.08	1,095.19
Net power (MW)	558.27	709.70	170.10	310.17	715.18	544.91
Total operation time (h)	193,200	185,700	262,800	193,200	185,700	193,200
Power ^{Exported} (MWh)	107,857,764	131,791,290	44,702,280	59,924,844	132,808,926	105,276,612
NPV (MMUSD)	2,060.68	-27.79	334.50	339.66	-478.73	1,144.21
CO ₂ ^{Captured} (MMt ^{CO2})	94.73	40.87	8.22	96.85	42.82	104.37
CO ₂ ^{Emitted} (MMt ^{CO2})	11.67	-	7.71	9.35	-	2.04
Power ^{Exported} /FCI (MWh/MMUSD)	180,699.57	153,134.40	57,591.50	68,507.46	109,300.43	96,126.34
NPV/Power ^{Exported} (USD/MWh)	19.1055	-0.2109	7.4828	5.6681	-3.6046	10.8686
NPV/CO ₂ Captured (USD/t ^{CO2})	21.7532	-0.6801	40.6931	3.5070	-11.1807	10.9628
NPV/CO ₂ ^{Emitted} (USD/t ^{CO2})	176.584	-	43.385	36.339	-	561.100
Power ^{Exported} /CO ₂ Emitted (MWh/t ^{CO2})	9.2426	-	5.7980	6.4111	-	51.6258
MAC (USD/t ^{CO2})	+4.58	+19.05	-4.30	+28.17	+20.97	+12.09

The results of Manzolini et al. (2013) did not perform well due to the negative *NPV*. The oxy-combustion NGCC comes second as it is technically a good solution, but its *NPV* is handicapped by high *FCI*.

 $NPV/CO_2^{Captured}$ (USD/ t^{CO2}) is similar to the previous ratio as it reports profitability per unit of environmental benefit, in which high values derive from high NPV and/or low environmental benefit (i.e., NGCC base cases attain infinite scores here). This is precisely the case of the post-combustion NGCC of Isogai and Nakagaki (2024), which attains the best score thanks to very low carbon capture. The post-combustion NGCC (this work) comes second due to its highest NPV, while the oxy-combustion NGCC comes third, doubly handicapped by its highest carbon capture and lower NPV.

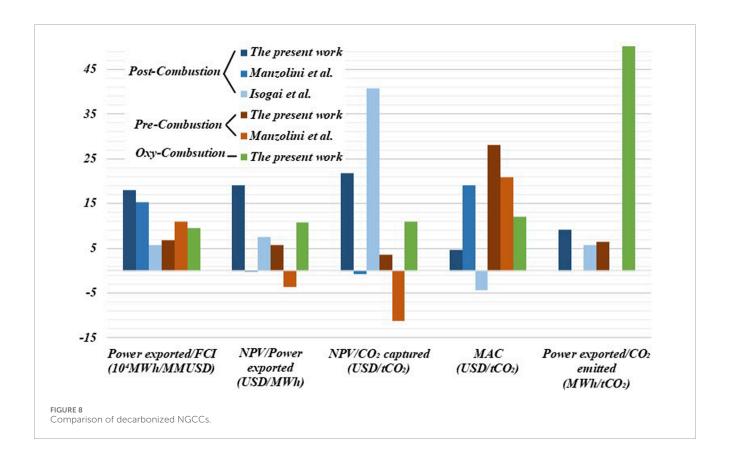
 $NPV/CO_2^{Emitted}$ (USD/t^{CO2}) is a metric reporting profitability per unit of environmental harm; that is, it favors plants that are doubly sustainable: both economically and environmentally. Here, the best plant with flying colors is the oxy-combustion NGCC, which is profitable and generates almost zero emissions. The post-combustion NGCC (this work) comes second, impelled by its highest NPV, which offsets its perceptible emissions, while the pre-combustion NGCCs are doubly handicapped by low NPV and perceptible emission load.

 $Power^{Exported}/CO_2^{Emitted}$ (MWh/t^{CO2}) is the best environmental sustainability indicator for power plants by excellence, as it reports

social benefit provided per unit of environmental harm. The oxycombustion NGCC attains, by far, the best score: more than 5.5× the post-combustion NGCC counterpart, which comes second, propelled by its high exported power. The pre-combustion NGCC comes third, handicapped by its lower power exportation and perceptible emission load.

Figure 8 presents a comparative graph illustrating MACs and several performance ratios for the decarbonized NGCCs from this work and from references. The decarbonized NGCCs from the present work achieved the highest NPV values, which entail high scores in NPV ratios. This is a consequence of considering CO_2 -to-EOR revenues in NGCCs with anti-carbon packages. The results of Manzolini et al. (2013) show the relevance of CO_2 pricing mechanisms, as demonstrated by their negative NPV and high MAC values for the post-combustion NGCC. These outcomes highlight that, without appropriate mechanisms such as carbon taxes or CCU, decarbonized power plants may remain financially unfeasible in the context of sustainable energy transition.

Figure 9 depicts the *MAC* distribution, also called the *MAC* curve (McKinsey and Company, 2007), for the six analyzed NGCCs with anti-carbon packages. This kind of graph typically ranks processes in ascending *MAC* order and defines the horizontal axis as cumulative lifetime carbon capture capacity. As already discussed,



the lowest and negative MAC corresponds to the post-combustion NGCC of Isogai and Nakagaki (2024), favored by its carbon tax penalty, which increased $AGAP^{25y}$ during decarbonization. The process with the lowest positive MAC is the post-combustion NGCC (this work) because it exhibits the lowest fall of $AGAP^{25y}$ (due to the lowest COM increase in Table 5) during decarbonization. In third place comes the oxy-combustion NGCC (164% MAC increase), handicapped by a greater $AGAP^{25y}$ decrease during decarbonization due to an impressive 100% COM increase, in response to a 240% FCI increase. Not surprisingly, the pre-combustion NGCC (this work) attains the highest MAC, doubly handicapped by the greatest loss of power exportation and 75% COM increase during decarbonization.

3.5 Sensitivity analysis

A sensitivity analysis was carried out to assess the influence of three independent factors (two economic factors and one operational factor) on the post-combustion NGCC, namely: (i) electricity price (P^{EL}) ranging from 80.5 USD/MWh to 98.5 USD/MWh, (ii) interest rate (i) ranging from 6% to 15%, and (iii) CO₂ capture efficiency (Cap^{EFF}) ranging from 85% to 94%. By sampling 10 equidistant levels for each factor, a total of 1,000 sensitivity points were obtained. The outputs included environmental and economic indicators: (i) NPV, (ii) MAC, (iii) FCI, (iv) total CO₂ captured ($CO_2^{CAP,25y}$), (v) electricity exported ($W^{EXP,25y}$) in the 25-year horizon, (vi) CO₂ emissions per year (CO_2^E), as well as performance ratios $NPV/CO_2^{E,25y}$ (economic return per environmental harm), $W^{EXP}/CO_2^{E,25y}$ (social benefit per

environmental harm), and *NPV/FCI* (economic return per USD invested).

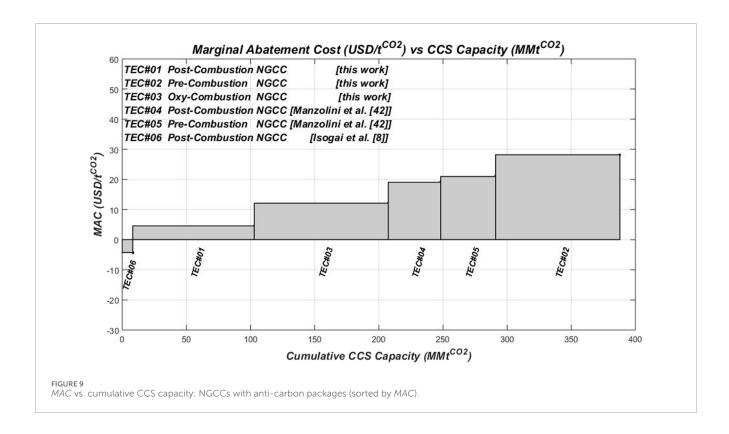
Other economic variables, necessary for calculating those outputs, are also presented: REV, COM, GAP, and AGAP. To estimate MAC in different economic scenarios, AGAP for the base case (zero capture) was recalculated, as it is influenced by both P^{EL} and i.

Table 7 highlights 12 selected sensitivity points that illustrate the general behavior, including the CCU-design point 205, and the base case (zero capture). Figures 10, 11 provide a visualization of all sensitivity points, where the color scale represents the NPV (Figure 10) and $NPV/CO_2^{E,25y}$ (Figure 11). A table with all 1,000 sensitivity points is available on request to the corresponding author.

The analysis revealed insights regarding the techno-economic performance of the post-combustion NGCC. First, all simulated cases achieved a positive *NPV*, indicating that *FCI* is fully recovered under the assumed operating and economic conditions for all sensitivity points (Figure 10). This attests to the economic resilience of the post-combustion NGCC even under pessimistic conjunctures.

Among the tested independent factors, the interest rate exerts the strongest influence on NPV (Figure 10). Its negative effect on NPV is illustrated by the paired points 1–91, 10–100, 901–991, and 910–1,000, which evince strong NPV decreases as i increases. By contrast, the capture efficiency has a minor influence on NPV; that is, NPV primarily responds to economic factors rather than CO_2 capture level (Figure 10).

GAP behavior reflects interactions between the price of electricity and capture efficiency. For a low electricity price, high capture efficiency entails greater *GAP*, unveiling financial



compensation of capture penalty losses with CO₂-to-EOR revenues. This behavior persists until $P^{EL}=90.5~USD/MWh$, where GAP becomes nearly insensitive to capture efficiency (points 501 and 510). Beyond this price, GAP is consistently higher for lower capture efficiency, indicating that CO_2 -to-EOR no longer compensates for electricity revenue losses.

For all economic scenarios, increasing capture efficiency reduces MAC. This occurs because $CO_2^{CAR,25y}$ increases, while AGAP remains comparatively unaffected. This trend is also seen in the $NPV/CO_2^{E,25y}$ and W^{EXP}/CO_2^{E} ratios, which increase as emissions decrease (capture efficiency increases). Overall, MAC, $NPV/CO_2^{E,25y}$, and W^{EXP}/CO_2^{E} improve as capture increases (Figure 11).

Finally, there is some trade-off between MAC and NPV. As NPV increases due to economic factors, MAC also increases because its numerator $\Delta AGAP^{25y}$ increases; that is, maximizing financial return entails a greater burden to the environment, unveiling the profitability vs. sustainability trade-off typical of CO_2 capture technologies.

4 Conclusion and outlook

Three NGCCs with anti-carbon packages were evaluated for sustainable fossil power production, namely, post-combustion NGCC, oxy-combustion NGCC, and pre-combustion NGCC. All decarbonized NGCCs export 100% of the captured CO_2 to EOR, generating extra revenues and ensuring profitability and economic sustainability. In consequence, the decarbonized NGCCs are all CCU power plants. The evaluation was based on the marginal abatement cost MAC (USD/t^{CO2}), classical

techno-economic outputs, such as $Power^{Exported-25y}$ (MWh), FCI (MMUSD), and NPV (MMUSD), and techno-environmental-economic performance ratios, including $Power^{Exported}/CO_2^{Emitted}$ (MWh/t^{CO2}) and $Power^{Exported-25y}/FCI$ (MWh/MMUSD).

All NGCC with anti-carbon packages have economic viability: (i) post-combustion NGCC $NPV^{25y} = 2,060.68$ MMUSD, $AGAP^{25y} = 3,794.49$ MMUSD; (ii) oxy-combustion NGCC: $NPV^{25y} = 1,144.21$ MMUSD, $AGAP^{25y} = 2,967.11$ MMUSD; and (iii) precombustion NGCC: $NPV^{25y} = 339.66$ MMUSD, $AGAP^{25y} = 1,499.79$ MMUSD.

The post-combustion NGCC has the lowest $MAC = +4.58~USD/t^{CO2}$ due to the lowest FCI and COM and the highest electricity export, entailing the highest $AGAP^{25y}$. Despite the highest FCI and COM, the oxy-combustion NGCC comes second with $MAC = +12.09~USD/t^{CO2}$ due to the greatest revenues after decarbonization. Meanwhile, the pre-combustion NGCC is characterized by the highest $MAC = +28.17~USD/t^{CO2}$, which is doubly handicapped by its low power export and high COM (from high FCI).

MAC values were also calculated for decarbonized NGCCs from the literature for comparison purposes. The post-combustion NGCC from Isogai and Nakagaki (2024) attained $MAC = -4.30~USD/t^{CO2}$, and the post-combustion NGCC and the precombustion NGCC from Manzolini et al. (2013) attained $MAC = 19.05~USD/t^{CO2}$ and $MAC = 20.97~USD/t^{CO2}$, respectively.

These *MAC* values lead to the general conclusion that post-combustion NGCCs operate with the lowest abatement cost, while pre-combustion NGCCs present the highest abatement cost. The oxy-combustion NGCC operates with intermediary abatement costs. The characteristic signature of post-combustion NGCC is the lowest investment and increase of revenues in decarbonized mode,

0.94

16.51

3.45 3.85 3.72 4.32 4.16 1.44 1.37

16.51

6.93

6.93

16.51

6.93

NPV/ FCI

TABLE 7 Sensitivity analysis selected points.

JSD/ USD

8.03

1.52

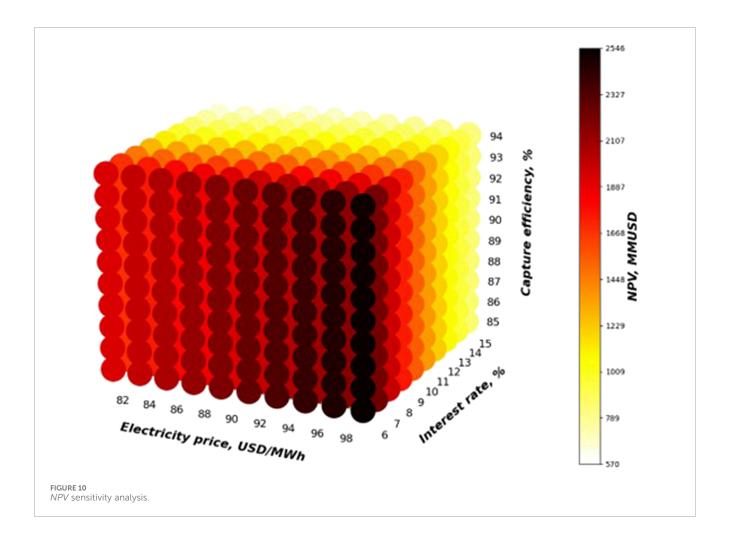
3.26

3.18 0.98

16.51

131.05 NPV/ CO₂^E 120.69 303.53 142.45 355.41 159.85 396.92 36.15 176.6 53.46 24 107.86 162.24 104.8104.8 104.8104.8 104.8 110.3 110.3 550.78 82.44 32.85 82.44 32.85 32.85 32.85 32.85 82.44 82.44 82.44 60.4 CO₂CAP,25y 100.00 100.00 100.00 100.00 100.00 90.48 90.48 90.48 90.48 90.48 94.73 0 4,622.48 AGAP W/ Ocap 4,228.63 1,701.22 22 4,228.63 5,240.41 10⁶ USD 1,701... 4,622. AGAP 1,591.47 4,228.63 3,601.69 3,794.5 4,101.36 4,100.64 1997.89 10^6 USD 4,521 GAP 386.18 10⁶ USD/ y 328.92 328.92 374.55 374.49 410.94 326.6 346.5 326.6 COM 342.43 342.43 342.43 342.43 342.43 10⁶ USD/ y 216.24 334.84 334.84 338.23 334.84 334.84 334.84 REV 10⁶ USD/ y 747.8 753.4 684.8 661.4 671.4 661.4 747.8 602.4 671.4 709.4 753.4 81.909 606.18 596.89 10⁶ USD 589.31 589.31 589.31 589.31 589.31 FCI 4.58 7.95 3.03 1.34 5.22 3.51 3.27 I.I7.4 2,546.12 2,519.28 1922.27 1926.53 2,268.85 2,255.84 NPV 10⁶ USD 2,554.33 575.79 569.85 2,060.7 851.46 831.76 94 0 85 94 85 94 89 85 94 85 94 85 15 15 15 15 9 9 9 9 9 9 9 9 USD/ MWh 84.5 80.5 84.5 90.5 98.5 98.5 80.5 80.5 80.5 90.5 98.5 98.5 1,000100 205** 510 910 ,0 10 16 501 106 166

*Base case. **CCU design point.

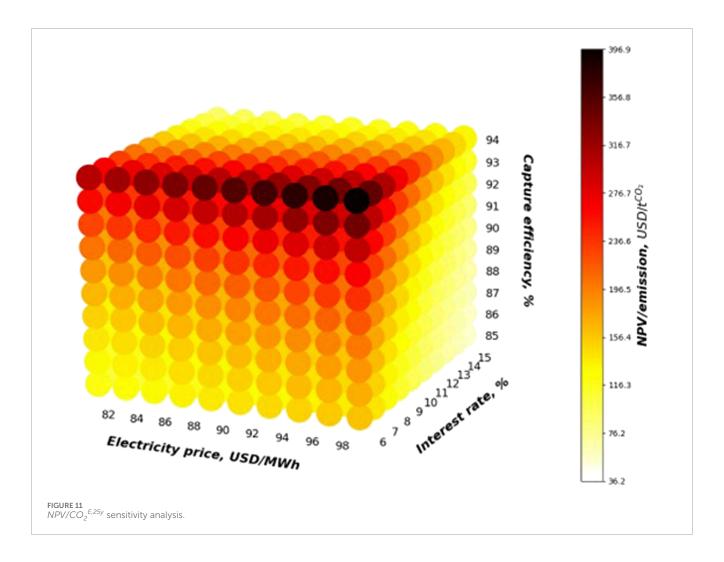


thanks to CO_2 monetization and low loss of electricity exportation. Meanwhile, the characteristic signature of pre-combustion NGCC comprises high investment and appreciable loss of electricity exportation after decarbonization. The oxy-combustion NGCC behaves in an intermediary way, with a characteristic signature consisting of the highest investment simultaneously with the highest revenues due to the highest CO_2 monetization and low loss of electricity exportation.

The analysis based on performance ratios corroborates the MAC analysis while unveiling novel facets of decarbonized NGCCs, namely, (i) Power Exported /FCI (MWh/MMUSD) unveils the social efficiency (social benefit per USD invested) of decarbonized NGCCs and indicates that the best sustainable NGCC is the post-combustion NGCC corroborating the aphorism, "with low money, prioritize post-combustion NGCC;" (ii) NPV/Power^{Exported} (USD/MWh) reports profitability per unit of social benefit, confirming postcombustion NGCC as the best option thanks to its highest NPV, while the oxy-combustion NGCC comes second as it is technically a good solution, but with a NPV handicapped by the highest FCI; (iii) NPV/CO₂^{Captured} (USD/t^{CO2}) reports profitability per unit of environmental benefit, which is the case of the postcombustion NGCC of Isogai and Nakagaki (2024), which attains the best score thanks to very low carbon capture, while the postcombustion NGCC (this work) comes second due to its highest NPV; (iv) NPV/CO₂^{Emitted} (USD/t^{CO2}) reports profitability per unit of environmental harm favoring plants that are economically and environmentally sustainable, indicating the oxy-combustion NGCC as the best decarbonized NGCC; the post-combustion NGCC (this work) comes second propelled by its highest NPV, while the pre-combustion NGCCs is doubly handicapped by low NPV and emission load; (v) $Power^{Exported}/CO_2^{Emitted}$ (MWh/t^{CO2}) is the environmental sustainability indicator for power plants by excellence that reports social benefit per unit of environmental harm, indicating the oxy-combustion NGCC as the best scorer, and the post-combustion NGCC comes second propelled by its high exported power, while the pre-combustion NGCC comes third, handicapped by its low power exportation and emission load.

Considering all metrics, the post-combustion NGCC is indicated as the best economic-environmental compromise, but the oxy-combustion NGCC comes close as the best social benefit per environmental harm. The pre-combustion NGCC is, in principle, not recommended as a decarbonized NGCC.

Sensitivity analysis confirms that the post-combustion NGCC, as the best short-term CCUS solution, is economically resilient; that is, it keeps a positive NPV under unfavorable economic scenarios. In addition, under low electricity prices, high capture efficiencies can even enhance economic performance due to CO_2 -to-EOR revenues. Another disclosure is that economic-environmental ratios like MAC, $NPV/CO_2^{E,25y}$, and W^{EXP}/CO_2^{E} all improve as capture increases, despite worse economic performance.



5 Limitations and future work

This work has scope, method, and results limitations; thus, suggestions for future works that eliminate part of such limitations apply, namely, (i) to consider catalyst deactivation costs in the precombustion NGCC; (ii) to consider uncertainties of CO_2 transport and logistics; (iii) to implement Monte Carlo analyses of results; (iv) different chemical or physical solvents for CO_2 absorption in post-combustion and pre-combustion NGCCs; (v) membrane permeation and/or gas-liquid membrane contactors for CO_2 capture instead of CO_2 absorption columns in post-combustion and pre-combustion NGCCs; (vi) different cryogenic air separation units (ASU) in an oxy-combustion NGCC; (vii) lifecycle analysis (LCA) of the three CCUS-NGCCs for complete evaluation of environmental aspects; (viii) a technology roadmap for short-term, mid-term, or long-term viability assessment of CCUS-NGCCs; and (ix) high-scale CCUS-NGCC optimization.

Data availability statement

The datasets presented in this article are not readily available because the funder does not allow distribution of information belonging to this research. Requests to access the datasets should be directed to José Luiz de Medeiros, jlm@eq.ufrj.br.

Author contributions

JG: Visualization, Validation, Formal Analysis, Data curation, Writing – original draft, Software, Investigation. Gd: Validation, Investigation, Visualization, Formal Analysis, Writing – original draft, Data curation, Software. JM: Investigation, Project administration, Conceptualization, Funding acquisition, Methodology, Writing – original draft, Supervision, Formal Analysis, Resources, Visualization, Writing – review and editing, Software. OA: Resources, Funding acquisition, Writing – review and editing, Project administration, Supervision.

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ensure accuracy, including review by the authors wherever possible. If you identify any issues, please contact us.

Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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Supplementary material

The Supplementary Material for this article can be found online at: https://www.frontiersin.org/articles/10.3389/fmats.2025.1679722/full#supplementary-material

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Nomenclature

AGAP Accumulated discounted gross annual profit (MMUSD)

AP Annual profit (MMUSD/y)

 C_{BM} , C_{BM}^o Bare module cost and bare module cost at the reference

conditions (MMUSD)

C_{GR} Grassroots module cost (MMUSD)

 $\mathbf{C_i}$ Concentration of component i (kmol/m³)

CEPCI CF Chemical engineering plant cost index (-), equipment

capacity factor (m², m³, kW, etc.)

COM, COL Cost of manufacture, cost of labor (MMUSD/y)

 ${\bf CUT, CRM, CCT} \qquad {\bf Cost} \quad {\bf of} \quad {\bf utilities,} \quad {\bf cost} \quad {\bf of} \quad {\bf raw} \quad {\bf materials,} \quad {\bf cost} \quad {\bf of} \quad {\bf CO}_2$

transportation (MMUSD/y)

DEPR Annual depreciation (MMUSD/y)

 ${f E_a}$ Activation energy (kJ/mol)

FCI, GAP Fixed capital investment (MMUSD), gross annual

profit (MMUSD/y)

i, ITR Annual interest and income tax rates (%)

 K_e , k_{H2O} Equilibrium constant, dissociative adsorption constant of H_2O

 ${f k_i}$ Absorption constant of chemical species i MAC Marginal abatement cost (USD/t^{CO2})

NH, N Project horizon (y)

NEQ, N_{OL} Number of equipment items, number of operators

NPV Net present value (MMUSD)

OP_{COST} Annual operator cost (MMUSD/y)

 P, P_i Pressure (bar), partial pressure of component i (bar) P, P_i Ideal gas constant (J/(mol.K)), reaction rate (mol/(s.g_{cat}))

REV Revenues (MMUSD/y)

T Temperature (K)

Greek symbols

 β Reversibility factor

Superscripts

ads Adsorption

Abbreviations

CCS Carbon capture and sequestration
CCU Carbon capture and utilization

CW Cooling water

DCC Direct-contact column

EGR Exhaust gas recirculation

EOR Enhanced oil recovery

IGCC Integrated gasification combined cycle

LPS Saturated low-pressure steam

MEA Monoethanolamine

MDEA Methyl-diethanolamine

NG Natural gas

NGCC NG combined cycle

PCC-MEA Post-combustion carbon capture with aqueous MEA

SMR Steam-methane reforming

TEG Triethylene-glycol

TSA Temperature-swing adsorption

WGS Water-gas shift.